Regioselectivity in the 1,3-Dipolar Cycloaddition Reaction of Unsymmetric Pyridinium Dicyanomethylides with Dimethyl Acetylenedicarboxylate and Methyl Propiolate: An Example of Dipole-Dipole Control of Regioselectivity?

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Dedicated to Professor Norman H. Cromwell

A study of the cycloaddition behavior of a series of unsymmetric pyridinium dicyanomethylides with dimethyl acetylenedicarboxylate and methyl propiolate has been carried out. The 1,3-dipolar cycloaddition proceeds in good yield with high regioselectivity to produce the corresponding indolizines and 1:1 adducts. The reactions of isoquinolinium dicyanomethylide follow frontier orbital predictions. In contrast, polar 3-substituted pyridinium dicyanomethylides gave predominantly the corresponding 8-isomers regardless of the substituents. The results can be explained by dipole-dipole interactions.

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1,3-Dipolar cycloaddition of heteroaromatic N-ylides [1] with activated alkenes and alkynes provides a convenient route to novel nitrogen bridged heterocycles such as indolizines [2-4], quinolizines [3], [2.2.3]cyclazines [5], mono-[6] and dibenzo[2.2.3]cyclazines [7]. Although stereochemical aspect of this reaction has been recently investigated systematically [8], only a few reports have been devoted to the regiochemical study of the reaction, neither of them being intensive nor extensive, especially regarding the ylide side regiochemistry [9,10].

We now briefly report on the regiochemical study of

1,3-dipolar reactions of unsymmetrically substituted pyridinium dicyanomethylides 1 with dimethyl acetylene-dicarboxylate (2a) and methyl propiolate (2b).

The reactions were performed in acetonitrile in a sealed tube at $40-50^{\circ}$ for 20-40 hours. The isomeric products were separated by medium-pressure liquid chromatography using a column (25×310 mm) prepacked with siliga gel (Lobar, LiChroprep Si 60, Merck). The regiochemical assignment of **3** and **4** is made on the basis of their 'H nmr spectra [11]. The isomeric ratios were determined by integration of appropriate signals ('H nmr) of the crude pro-

ducts. The results are summarized in Table 1. The reaction of 3-substituted pyridinium dicyanomethylides 1 with

Table 1
Reactions of Ylides 1 with Alkynes 2

Entry No.	Ylide		Alkyne Total Yield		Ratio	
·	R¹	R²	X	%	3	4
1	Me	Н	CO₂Me	40	76	24 [a]
2	CO ₂ Me	Н	CO_2Me	61	73	27
3	CN	Н	CO_2Me	96	80	20
4	COMe	Н	CO_2Me	57	100	0
5	$-(CH = CH)_2$		CO₂Me	61	100	0
6	Me	Me	H	100	62	38
7	Me	Н	Н	80	66	34
8	CO₂Me	Н	H	63 [b]	100	0
9	CN	H	Н	62 [b]	100	0
10	COMe	Н	Н	77 [b]	100	0
11	$-(CH = CH)_2$		Н	94	100	0
12	$-(CH = CH)_2-$		Ph	99	94	6

[[]a] Taken from ref [9]. [b] The product was 1:1 adduct 5.

2a in acetonitrile at ca. 55° took place smoothly to give the corresponding indolizines 3 and 4, 3 being predominantly formed in all the cases examined (entries 1-4). Particularly, 3-acetylpyridinium dicyanomethylide (1e) produced the 8-acetyl isomer 3ea exclusively.

Methyl propiolate (2b) that is generally known as a less reactive dipolarophile than 2a also reacted smoothly with 3-substituted pyridinium dicyanomethylides 1 to give either the corresponding indolizines 3 and 4 or 1:1 adducts 5 depending on the nature of the substituents. Interaction of pyridinium ylides 1c-e possessing an electron withdrawing group with 2b gave rise to a single 1:1 adduct 5cb-eb. These adducts were converted to the corresponding indolizines 3cb-eb upon treatment with Pd/C in refluxing toluene. The regiochemical assignment with respect to 1,2-position rest on their 'H nmr spectra [12]. Isoquinolinium dicyanomethylide (1f) with 2a and 2b gave regiospecifically 3fa and 3fb respectively (entries 5 and 11).

It is now well documented that, to a more or less extent, most 1,3-dipolar cycloadditions receive contribution from HOMO-LUMO interaction [1,13]. The results of CNDO/2 calculations [14] both for the ylides and alkynes are depict-

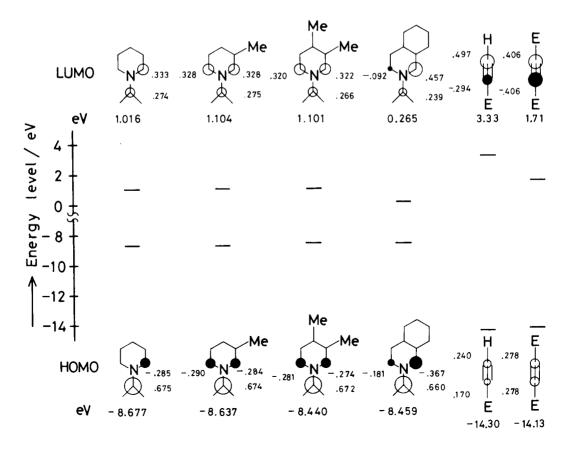


Figure 1. Frontier molecular orbital interaction in 1,3-dipolar cycloaddition of 1a,b,f with 2a and 2b.

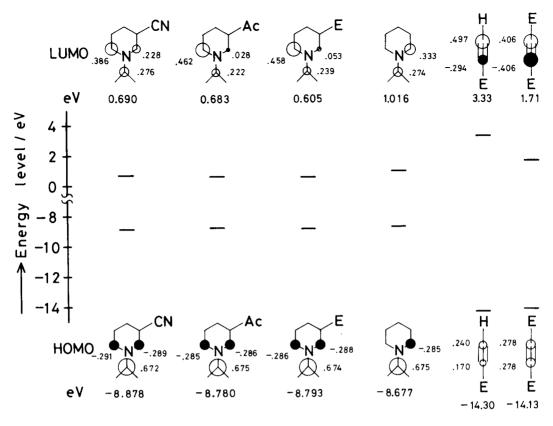


Figure 2. Frontier molecular orbital interaction in 1,3-dipolar cycloaddition of 1b-d with 2a and 2b.

ed in Figures 1 and 2, an inspection of which clearly shows that the present reaction is predominantly HOMO(1,3-dipole)-LUMO(dipolarophile) controlled. Indeed, the regiochemical consequence with respect to 2b is in agreement with the FMO interactions. The exclusive formation of 3fa and 3fb is also in accord with such considerations. The slightly reduced regioselectivity in the reaction of 1f with 2c is likely steric (by phenyl) in origin (entry 12). However, the coefficients of 2- and 6-positions of polar 3-substituted pyridinium dicyanomethylides 1a-e are almost of the same magnitude. This does not explain the observed regioselectivity, that is, predominant formation of the 8-isomers 3.

Since the answer to this regiochemical control does not seem (at present) to be attributable to FMO factors, we examined simple dipole-dipole interactions [1], as depicted in Figure 3. The interaction of the apparently more dipolar azomethine ylides 1c-e with 2 will afford predominantly 3, while the less dipolar azomethine ylides 1a,b would result in less regioselectivity. We assume here the dipole is induced more efficiently by 3-substituent through the 2-position rather than the 6-position. The regiochemical outcome with respect to 2b is also in agreement with dipole-dipole interaction. Thus, regiochemical control in the cycloaddition of polar 3-substituted ylides with 2a,b appears to be dipole-dipole interactions rather than FMO interactions. Recognition of the importance of dipole-dipole

factors in controlling the regiochemistry of ylide cycloaddition sets the stage for further studies with related systems.

Figure 3. Dipole-dipole interaction in the 1,3-dipolar cycloaddition of polar 3-substituted pyridinium dicyanomethylides with methyl propilate.

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- [11] E.g., ¹H nmr (deuteriochloroform, 90 MHz); **3da**, 3.97 (s, 6H, $CH_3O \times 2$), 7.07 (bt, J = 7.1 Hz, 1H, H-6), 7.66 (dd, J = 7.1, 1.1 Hz, 1H, H-7), 8.45 (dd, J = 7.0, 1.1 Hz, 1H, H-5); **4da**, 3.90, 3.98 (each s, 3H $\times 2$, $CH_3O \times 2$), 7.34 (bd, J = 9.0 Hz, 1H, H-7), 8.30 (bd, J = 9.0 Hz, 1H, H-8), 8.63 (bs, 1H, H-5).
- [12] E.g., ¹H nmr (deuteriochloroform, 90 MHz): **3ab**, 2.76 (s, 3H, CH₃), 3.83 (s, 3H, CH₃O), 6.90 (q, J = 7.4, 7.5 Hz, 1H, H-6), 7.02 (bd, J = 7.5 Hz, H-7), 7.73 (s, 1H, H-2), 8.16 (bd, J = 7.4 Hz, H-5); **4ab**, 2.38 (s, 3H, CH₃), 3.87 (s, 3H, CH₃O), 7.13 (dd, J = 10, 1.5 Hz, H-7), 7.66 (s, 1H, H-2), 8.10 (bs, 1H, H-5), 8.13 (d, J = 10 Hz, H-8).
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